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## Structure Reports

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**(2Z)-1-(5-Hydroxy-3-methyl-1-phenyl-1H-pyrazol-4-yl)-3-(4-methoxyanilino)-but-2-en-1-one**Abdullah M. Asiri,<sup>a,b,‡</sup> Abdulrahman O. Al-Youbi,<sup>a</sup> Hassan M. Faidallah,<sup>a</sup> Seik Weng Ng<sup>c,a</sup> and Edward R. T. Tiekink<sup>c\*</sup><sup>a</sup>Chemistry Department, Faculty of Science, King Abdulaziz University, PO Box 80203, Jeddah, Saudi Arabia, <sup>b</sup>The Center of Excellence for Advanced Materials Research, King Abdulaziz University, Jeddah, PO Box 80203, Saudi Arabia, and <sup>c</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia  
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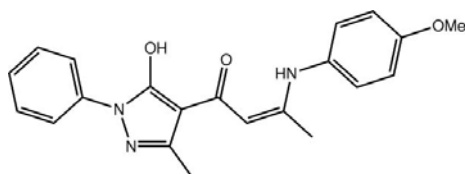
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Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.042;  $wR$  factor = 0.104; data-to-parameter ratio = 15.4.

The central residue in the title compound,  $\text{C}_{21}\text{H}_{21}\text{N}_3\text{O}_3$ , is close to planar (r.m.s. deviation = 0.0753 Å for all non-H atoms from OH to NH inclusive): the hydroxy, amino and carbonyl groups all lie to the same side of the molecule (the conformation about the ethene bond is *Z*), facilitating the formation of intramolecular  $\text{O}-\text{H}\cdots\text{O}$  and  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds that close *S*(6) rings. However, overall the molecule is twisted as the terminal aromatic rings are not coplanar with the central plane [dihedral angles = 20.55 (5) and 80.90 (4)° for the N-bound phenyl ring and the methoxybenzene ring, respectively]. The dihedral angle between the rings is 82.14 (7)°. Supramolecular layers in the *ac* plane mediated by  $\text{C}-\text{H}\cdots\pi$  interactions are found in the crystal.

## Related literature

For background to the synthesis, see: Gelin *et al.* (1983); Bendaas *et al.* (1999). For the structure of the 4-chloro derivative, see: Asiri *et al.* (2011).



## Experimental

## Crystal data

 $\text{C}_{21}\text{H}_{21}\text{N}_3\text{O}_3$   
 $M_r = 363.41$ Monoclinic,  $P2_1/n$   
 $a = 9.5717$  (3) Å<sup>‡</sup> Additional correspondence author, e-mail: aasiri2@kau.edu.sa. $b = 16.9516$  (6) Å  
 $c = 11.3143$  (4) Å  
 $\beta = 104.946$  (4)°  
 $V = 1773.70$  (10) Å<sup>3</sup>  
 $Z = 4$ Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 100$  K  
 $0.30 \times 0.25 \times 0.20$  mm

## Data collection

Agilent SuperNova Dual  
diffractometer with an Atlas  
detector  
Absorption correction: multi-scan  
(*CrysAlis PRO*; Agilent, 2010)  
 $T_{\min} = 0.837$ ,  $T_{\max} = 1.000$ 8486 measured reflections  
3939 independent reflections  
3145 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.024$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.104$   
 $S = 1.05$   
3939 reflections  
255 parameters  
2 restraintsH atoms treated by a mixture of  
independent and constrained  
refinement  
 $\Delta\rho_{\max} = 0.27$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.23$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

 $\text{Cg1}$  and  $\text{Cg2}$  are the centroids of the  $\text{N1,N2,C1-C3}$  and  $\text{C15-C20}$  rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1}\cdots\text{O2}$	0.86 (1)	1.68 (1)	2.4963 (15)	156 (2)
$\text{N3}-\text{H3}\cdots\text{O2}$	0.89 (1)	1.92 (1)	2.6447 (16)	138 (2)
$\text{C14}-\text{H14b}\cdots\text{Cg1}^{\text{i}}$	0.98	2.88	3.5542 (18)	127
$\text{C21}-\text{H21c}\cdots\text{Cg2}^{\text{ii}}$	0.98	2.76	3.5195 (17)	134

Symmetry codes: (i)  $-x + 1, -y + 1, -z + 1$ ; (ii)  $-x + 2, -y + 1, -z$ .

Data collection: *CrysAlis PRO* (Agilent, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6355).

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